UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the May/June 2008 question paper

5129 COMBINED SCIENCE

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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1	(a)	(i) tree / grass / flower	[1]
		(ii) cow / snail / rabbit	[1]
	(b)	Box 2 = snail Box 4 = hawk	[2]
	(c)	Sun / sunlight (not light alone)	[1]
	(d)	decomposer / bacteria / fungi	[1]
	(e)	energy supply is limited energy is used by the organisms energy is lost at each stage insufficient energy left (for another level)	[2]
2	(a)	copper / Cu	[1]
	(b)	potassium / K	[1]
	(c)	iron / Fe	[1]
	(d)	copper / Cu	[1]
	(e)	zinc / Zn	[1]
3	(a)	(i) Q = It or 0.2 × 180 36 (0.6 gains 1 mark)	[2]
		(ii) $V = IR \text{ or } 7 \times 0.2$ = 1.4	[2]
	(b)	0.6 / 2.0 –(a)(ii)	[1]

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Syllabus 5129 Paper 02

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4 (a) limewater milky / cloudy / white (precipitate)

[2]

(b) (i)
$$CH_4 = 16$$

 $CO_2 = 44$ [2]

(ii) 16 → 44 ∴ 4 → 44 × 4/16 = 11 g correct method from wrong numbers in (b)(ii) gains 2

[2]

5 (a) blue pink

[1]

(b) (i) transpiration

[1]

(ii) upper surface has waxy layer fewer / no stomata answer could be in terms of lower surface

[2]

(c) root hair osmosis

[2]

6 (a) reduction

[1]

(b) conducts electricity conducts heat malleable ductile high density high melting point high boiling point

any 2

(c) boils at 100 °C/boils at single temperature

[1]

[2]

7 (a) (i) gravity / weight

[1]

(ii) gravitational / potential

[1]

(b) line is curved / not straight

[1]

(c)
$$F = ma$$
 or $a = F/m$ or $300/80$
= 3.75
 m/s^2

[3]

	Page 4		Mark Scheme	Syllabus	Paper	
			GCE O LEVEL – May/June 2008	5129	02	
8	(a)	matt bla	ack is a better absorber / shiny is a better reflector		[1]	
	(b)	(i) sta	ys the same / no change / none		[1]	
		(ii) ded	creases / gets less / lowers		[1]	
	(c)	microw	aves and radiowaves (either order)		[2]	
9	(a)	large pi mixes f	rinds food ecces to smaller pieces ood with saliva es (soluble particles) any 2		[2]	
	(b)	secrete lubricat	liquid / saliva enzymes / amylase e / softens food es convert starch to maltose / sugar	ny 2	[2]	
	(c)	bacteria cavities	a / enamel to dissolve / tooth decay		[2]	
10	(a)	hydroge	en / H ⁺		[1]	
	(b)	(i) red			[1]	
		(ii) ora	nge / yellow		[1]	
	(c)	(i) Mg	+ $H_2SO_4 \rightarrow MgSO_4 + H_2$		[1]	
		ma	gnesium carbonate gnesium hydroxide gnesium oxide		[2]	
11	(a)	like cha	rges (repel)		[1]	
	(b)	positive			[1]	

Mark Scheme

Syllabus

Paper

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Page 5			Mark S	Scheme		Syllabus	Paper
			GCE O LEVEL – May/June 2008			5129	02
12		quency = I iod = s	Hz or s ⁻¹				[2]
13	(a)	and one	showing 3 bonding pairs lone pair shell drawn it must be corre	ect)			[2]
	(b)	400–500 200–300 iron			[3]		
	(c)	potassiui phospho	m orus (either order)				[2]
14	(a)	lack of (e	enough) food				[1]
	(b)	not enou too much civil unre earthqua	h rain / water / floods	rks	plants die / ci no photosynt plants washe no one to ten or crops dest food destroye	hesis / growth d away / die d crops royed	[4]
15	(a)	both pos	al shape with sitive and negative values cycles shown				[3]
	(b)	stronger	ed speed of rotation magnet rns in coil		any 1		[1]
16	(a)	0.1 × 30 W = 0.1	= W × 0.2 15				[2]
	(b)		clockwise / iron rod goes do attracted by the magnet	own / left goes	s down		[2]

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17	(a)	(i) ele	ectro	onic stru	cture dr	awn as 2	2 8					[1]
		(ii) +3	•									[1]
	(b)	group 3 forms a positive ion on the left of the Periodic Table trend across period is metal to non-metallic any 2									[2]	
	(c)	protected by a layer of (aluminium) oxide OR oxide layer / on surface of metal										[2]
18	(a)	A = tes B = cot C = plu D = rac	tyled ımul	le								[4]
	(b)	water oxyger suitable		named te	emperat	ure						[3]
19	(a)	26 – 14 (one co			g from	diagrams	s gains 1	mark)				[2]
	(b)	0.24 o	r (a	a) /50								[1]
20	(a)	5										[1]
	(b)	extens load =	ion : 4	= 10 (cn	1)							[2]